Selective Skeletal Rearrangement by Carbon–Carbon Bond Activation

Tamotsu Takahashi,*a Takayasu Fujimori,a Takashi Seki,a Masahiko Saburi,a Yasuzo Uchida,a Christophe J. Rousset,^b and Ei-ichi Negishi*^b

^a Department of Industrial Chemistry, Faculty of Engineering, The University of Tokyo, 7-3-1 Hongo, Bunkyo-ku, Tokyo 113, Japan

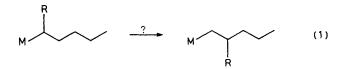
^b Department of Chemistry, Purdue University, West Lafayette, IN 47907, U.S.A.

Skeletal rearrangement of α -substituted organometallic compounds to give β -substituted compounds by carbon–carbon bond activation was performed with bis(η^5 -cyclopentadienyl)-2,5-dimethyl-zircona- or -hafna-cyclopentane, which were regio- and stereo-selectively converted into the corresponding 3,4-dimethylmetallacyclopentanes in high yield.

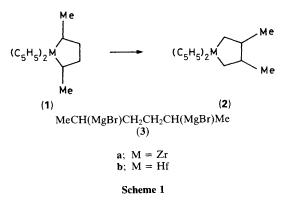
Selective skeletal rearrangement by carbon–carbon bond activation is a potentially attractive methodology in organic synthesis. Reaction (1) for organometallic compounds can be considered to be representative. We now describe a novel selective rearrangement by carbon–carbon bond activation using organozirconium or organohafnium compounds (Scheme 1).

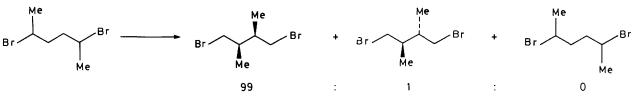
The zirconium complex (1a) with methyl groups on the α and α' -carbon atoms was prepared *in situ* from $(C_5H_5)_2ZrCl_2$ and the corresponding Grignard reagent (3), and underwent selective skeletal rearrangement during 1 h at room temperature by carbon-carbon activation to give 2,3-dimetylbutane after hydrolysis in 98% yield based on the metal. Bromination of the reaction mixture instead of hydrolysis gave (\pm) -1,4dibromo-2,3-dimethylbutane selectively $[(\pm):meso \ 99:1]$ (Scheme 2). 2,5-Dibromohexane was totally absent.

The hafnium derivative (1b) reacted similarly, at higher temperature. The rearrangement product 2,3-dimethylbutane was obtained in 53 and 98% yield during 1 h at 50 and 80 $^{\circ}$ C,

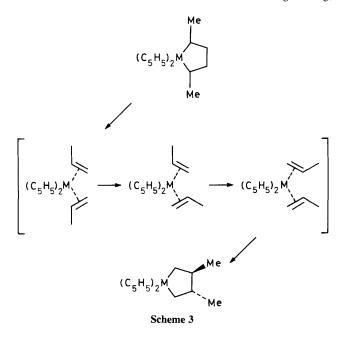


respectively, after hydrolysis. The reaction was monitored by ¹H NMR spectroscopy. Reaction of $(C_5H_5)_2$ HfCl₂ with the Grignard reagent (3) quantitatively afforded (1b), the ¹H NMR spectrum of which showed a C_5H_5 signal at δ 5.89. At 50 °C three new singlets assignable to the C_5H_5 groups of (2b) and two intermediates, 2,4-dimethylhafnacyclopentane isomers, emerged at δ 5.93, 5.90, and 5.87, respectively, and grew at the expense of that of (1b) at δ 5.89. This process obeyed first-order kinetics, $k_1 = (6.5 \pm 0.2) \times 10^{-3} \text{ min}^{-1}$ at 50 °C. The final spectrum revealed only one singlet peak for





Scheme 2. Reagents: Mg, then $(C_5H_5)_2$ ZrCl₂, then Br₂.



the C_5H_5 group of (2b) at δ 5.93, signals due to the C_5H_5 groups of the starting material and the two intermediates having completely disappeared.

Zircona- and hafna-cyclopentane complexes undergo carbon-carbon bond cleavage to give zirconium and hafnium ethylene complexes, 1–3 respectively. Similar transformations of some other metallacyclopentanes have been reported.⁴ As expected, addition of an excess of PMe₃ to bis(η^5 -cyclopentadienyl)dimethylzirconacyclopentane prepared *in situ* led to carbon-carbon bond cleavage to produce the zirconocenepropene complex which was identified by its characteristic ¹H, ¹³C, and ³¹P NMR spectra, which were the same as those of an authentic sample obtained from dipropylzirconocene complex.⁵

Furthermore, a cross-over experiment using $bis(\eta^5-$ cyclopentadienyl)zirconacyclopentane and (2a) gave 1,4-

dibromo-2-methylbutane in 10% yield after bromination of the mixture.

These results strongly suggest a β , γ -carbon–carbon bond activation of (1). Bis(alkene)complexes have been proposed as chemical species after β , γ -carbon–carbon bond fission of metallacyclopentane systems.^{3,4} Although the determination of the structure of the actual intermediate species of the rearrangement reactions must await further investigations, the skeletal rearrangements can be explained by the mechanism in Scheme 3 involving β , γ -carbon–carbon bond activation, followed by formation of a bis(alkene) complex, rotation of the alkenes due to steric hindrance, and selective coupling of two alkenes on the metal.

Recently we have found regio- and stereo-selective alkene coupling reactions on zirconium or hafnium compounds.⁶ The reaction of propene with $(C_5H_5)_2Zr(MeCH=CH_2)^5$ prepared *in situ* from $(C_5H_5)_2ZrPr_2$ afforded, after bromination, 1,4-dibromo-2,3-dimethylbutane, displaying a very similar regioand stereo-selectivity pattern as the skeletal rearranged product.

This research was supported by a Grant-in-Aid for Scientific Research No. 63470078 from the Ministry of Education, Science and Culture and the National Science Foundation (CHE-8842237).

Received, 25th September 1989; Com. 9/04067F

References

- 1 T. Takahashi, M. Tamura, M. Saburi, Y. Uchida, and E. Negishi, J. Chem. Soc., Chem. Commun., 1989, 852.
- 2 G. Erker, P. Czisch, C. Kruger, and J. Wallis, *Organometallics*, 1985, 4, 2059.
- 3 G. Erker, V. Dorf, and A. L. Rheingold, *Organometallics*, 1988, 7, 138.
- 4 R. H. Grubbs and A. Miyashita, J. Am. Chem. Soc., 1978, 100, 1300.
- 5 T. Takahashi, M. Murakami, M. Kunishige, M. Saburi, Y. Uchida, K. Kozawa, T. Uchida, D. R. Swanson, and E. Negishi, *Chem. Lett.*, 1989, 761.
- 6 D. R. Swanson, C. J. Rousset, E. Negishi, T. Takahashi, T. Seki, M. Saburi, and Y. Uchida, J. Org. Chem., 1989, 54, 3521.